



INDIAN SCHOOL SOHAR

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CBSE Affiliation No.: 6130006



TRANSFER CERTIFICATE

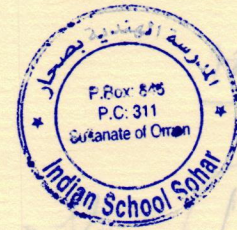
Affiliation No. 6130006

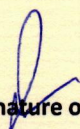
Sl.No. 2184


School Code. 90169


Admission No. 14582

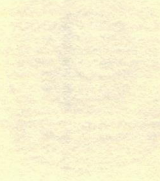
1. Name of Pupil : AISHVI AVESH
2. Mother's Name : AMRUTHA
3. Father's / Guardian's Name : AVESH
4. Date of Birth (in Christian Era) according to Admission and Withdrawal Register : (in figures): 02/02/2017
(in words): TWO FEBRUARY TWO THOUSAND SEVENTEEN
5. Nationality : INDIAN
6. Whether the candidate belongs to Schedule Caste or Schedule Tribe or OBC : -----
7. Date of first admission in the school with class : 07-05-2023 - Class I
8. Class in which the pupil last studied : (in figures) : I (in words): FIRST
9. School / Board Annual examination last taken with result : LEFT WHILE STUDYING IN CLASS I
10. Whether failed, if so once / twice in the same class: : N/A
11. Subjects Studied : ENGLISH, EVS, HINDI, MATHEMATICS
12. Whether qualified for promotion to the higher class : -----
(in figures) ----- (in words) : -----
13. Month upto which the pupil has paid school dues : MAY-2023
14. Any fee concession availed of. If so, the nature of such concession : -----
15. Total No. of working days in the academic session : 39
16. Total No. of working days pupil present in the school : 19
17. Whether NCC Cadet / Boy Scout / Girl Guide (details may be given) : NA
18. Games played or extra curricular activities in which the pupil usually took part: : -----
(mention achievement level therein)
19. General Conduct : GOOD
20. Date of application for certificate : 31/05/2023
21. Date of issue of certificate: : 31/05/2023
22. Reason for leaving the school : PARENT'S REQUEST
23. Any other remarks : NO




Signature of Class Teacher


Checked by
(with full name & designation)


Signature of Principal with date
School Seal



EXERCISES

Page No. _____
Date _____

1. Name the parts of the cell.
2. Write the chemical reaction taking place in the cell.
3. Explain the function of the salt bridge.

4. Why is a salt bridge necessary in a galvanic cell?
5. What is the role of the salt bridge in maintaining the electrical neutrality of the half-cells?

6. How does the concentration of the electrolyte affect the cell potential?
7. Write the Nernst equation for the cell.

8. Calculate the standard cell potential for the following cell:
Zn | Zn²⁺ (1M) || Cu²⁺ (1M) | Cu

9. Calculate the standard cell potential for the following cell:
Fe | Fe²⁺ (1M) || Ag⁺ (1M) | Ag

10. Calculate the standard cell potential for the following cell:
Ni | Ni²⁺ (1M) || Pb²⁺ (1M) | Pb

ANESH

Hareesh